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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/696,401	10/29/2003	Vladimir Grushin	PE0649USDIV4	6733
23906	7590	06/28/2005		
E I DU PONT DE NEMOURS AND COMPANY LEGAL PATENT RECORDS CENTER BARLEY MILL PLAZA 25/1128 4417 LANCASTER PIKE WILMINGTON, DE 19805			EXAMINER KIELIN, ERIK J	
			ART UNIT 2813	PAPER NUMBER
			DATE MAILED: 06/28/2005	

Please find below and/or attached an Office communication concerning this application or proceeding.

Office Action Summary	Application No.	Applicant(s)
	10/696,401	GRUSHIN ET AL. <i>(RM)</i>
	Examiner	Art Unit
	Erik Kielin	2813

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --
Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If the period for reply specified above is less than thirty (30) days, a reply within the statutory minimum of thirty (30) days will be considered timely.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133). Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

Status

1) Responsive to communication(s) filed on 29 October 2003.
 2a) This action is FINAL. 2b) This action is non-final.
 3) Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

Disposition of Claims

4) Claim(s) 12-15 is/are pending in the application.
 4a) Of the above claim(s) _____ is/are withdrawn from consideration.
 5) Claim(s) _____ is/are allowed.
 6) Claim(s) 12-15 is/are rejected.
 7) Claim(s) _____ is/are objected to.
 8) Claim(s) _____ are subject to restriction and/or election requirement.

Application Papers

9) The specification is objected to by the Examiner.
 10) The drawing(s) filed on 29 October 2003 is/are: a) accepted or b) objected to by the Examiner.
 Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).
 Replacement drawing sheet(s) including the correction is required if the drawing(s) is objected to. See 37 CFR 1.121(d).
 11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.

Priority under 35 U.S.C. § 119

12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).
 a) All b) Some * c) None of:
 1. Certified copies of the priority documents have been received.
 2. Certified copies of the priority documents have been received in Application No. _____.
 3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

* See the attached detailed Office action for a list of the certified copies not received.

Attachment(s)

1) <input checked="" type="checkbox"/> Notice of References Cited (PTO-892)	4) <input type="checkbox"/> Interview Summary (PTO-413)
2) <input type="checkbox"/> Notice of Draftsperson's Patent Drawing Review (PTO-948)	Paper No(s)/Mail Date. _____
3) <input checked="" type="checkbox"/> Information Disclosure Statement(s) (PTO-1449 or PTO/SB/08) Paper No(s)/Mail Date <u>10/29/03 10/13/04</u> .	5) <input type="checkbox"/> Notice of Informal Patent Application (PTO-152)
	6) <input type="checkbox"/> Other: _____

DETAILED ACTION

Information Disclosure Statement

The information disclosure statement filed 13 October 2004 fails to comply with the provisions of 37 CFR 1.97, 1.98 and MPEP § 609 because some of the references have not been provided with dates in accordance with 37 CFR 1.98(b)(5). Also the MPEP 609 states,

“Each publication must be identified by publisher, author (if any), title, relevant pages of the publication, and date and place of publication. The date of publication supplied must include at least the month and year of publication, except that the year of publication (without the month) will be accepted if the applicant points out in the information disclosure statement that the year of publication is sufficiently earlier than the effective U.S. filing date and any foreign priority date so that the particular month of publication is not in issue.” (Emphasis added.)

The IDS has been placed in the application file, but only the references initialed by Examiner have been considered. Applicant is advised that the date of any re-submission of any item of information contained in this information disclosure statement or the submission of any missing element(s) will be the date of submission for purposes of determining compliance with the requirements based on the time of filing the statement, including all certification requirements for statements under 37 CFR 1.97(e). See MPEP § 609 ¶ C(1).

Claim Rejections - 35 USC § 103

1. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

Art Unit: 2813

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negated by the manner in which the invention was made.

2. Claims 12-15 are rejected under 35 U.S.C. 103(a) as being unpatentable over US Patent Application Publication 2002/0182441 A1 (**Lamansky et al.**) in view of the article **Dedeian et al.** "A new synthetic route to the preparation of a series of strong photoreducing agents: *fac* tris-*o*-thro-metalated complexes of iridium(III) with substituted 2-phenylpyridines" Inorganic Chemistry, Vol. 30, 1991, 1685-1687 and US Patent Application Publication 2002/0048689 A1 (**Igarashi et al.**).

Regarding claim 12, **Lamansky** discloses iridium (Ir) metal complexes for use in electroluminescent devices having (1) two bidentate ligands including, *inter alia*, fluorine-substituted 2-phenylpyridines, and two monodentate ligands including, *inter alia*, those shown in Figs. 6(b) and 6(c), and include chlorine. Exemplary complexes have the two bidentate and two monodentate ligands are shown in Figs. 8(b) and 8(c). Each of the phenyl and pyridine rings of the 2-phenylpyridines are shown more generally to be substituted by as shown in Figs. 5a and 5b. The substituents are, *inter alia*, halogen (e.g. fluorine) and alkyl.

Lamansky states in paragraph [0049],

"The at least one mono-anionic, bidentate, carbon-coordination ligand of the present invention is substituted with electron donating and/or electron withdrawing substituents that shift the emission, relative to the un-substituted ligand, to either the blue, green or red region of the visible spectrum. The particular substituents used on particular ligands will depend upon the desired shift in emission. Generic representative examples of the at least one mono-anionic, bidentate, carbon-coordination ligand of the present invention are listed in FIGS. 5a, 5b and 5c." (Emphasis added.)

Accordingly, **Lamansky** suggests using various halogen and alkyl substituents on the phenylpyridine ligands to modify the emission wavelength of the Ir metal complex. **Lamansky** does not however, specifically indicate that the alkyl is specifically trifluoromethyl on either of the 2-phenylpyridines or the phosphine (claim 13 only). Note however that **Lamansky** gives a specific example of the alkyl-substituted phenylpyridine as a methyl-substituted phenylpyridine in Fig. 5b.

Dedeian, like **Lamansky**, discloses electroluminescent phenylpyridine ligands for Ir. **Dedian** shows in the upper, right-hand corner of p. 1686, Table I, that phenyl pyridine can be substituted with either a methyl (as in one example in Lamansky) or trifluoromethyl.

It would have been obvious for one of ordinary skill in the art, at the time of the invention to use trifluoromethyl as the substituent on each of the 2-phenylpyridines because, as noted above, **Lamansky** suggests using various halogen and alkyl substituents to modify the emission wavelength of the Ir metal complex at any position on the ring, and **Dedian** teaches the trifluoromethyl is a known substituent for ligands of Ir metal complexes. Importantly, the **Lamansky** and **Dedian** references share a common author, Peter Djurovich, such that one of ordinary skill would also be aware that each group was aware the alkyl in **Lamansky** suggests fluoroalkyl substituents as well.

Lamansky does not teach carbon monoxide as one of the monodentate ligands for the Ir metal complex, but suggests others are known such as those indicated in Comprehensive Coordination Chemistry, Vols. 1-7, G. Wilkinson, Ed., Pergamon Press, 1987 (Lamansky, paragraphs [0055] and [0056]).

Igarashi teaches iridium (Ir) metal complexes for use in electroluminescent devices including the same general formula as that in **Lamansky**, specifically two, bidentate ligands, (specifically substituted phenyl-pyridines) and two monodentate ligands. **Igarashi** teaches that carbon monoxide is a known monodentate ligand for used in Ir metal complexes, citing the text Comprehensive Coordination Chemistry, the same text in which **Lamansky** suggests other ligands can be found (**Igarashi**, paragraph [0049] and p. 15, formula (1-77)).

It would have been obvious for one of ordinary skill in the art, at the time of the invention to use carbon monoxide as one of the monodentate ligands because **Lamansky**, as noted above, suggests modifying the ligands to shift the emission wavelength and also suggests using known ligands in the same text as **Igarashi** suggests, one of which is carbon monoxide.

Regarding claims 13 and 14, **Lamansky** discloses that the iridium complexes are used in the light-emitting layer of an electronic, light-emitting device (Figs. 18(a) and (b), paragraph [0013], [0128], and [0129]).

Regarding claim 15, because the light-emitting layer is the location where electrons and holes charges are transported to recombine with the emission of light. The light-emitting layer is a charge transport layer because the charges must move through the light-emitting layer in order to recombine.

Conclusion

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Erik Kielin whose telephone number is 571-272-1693. The examiner can normally be reached from 9:00 - 19:30.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Carl Whitehead, Jr. can be reached on 571-272-1702. The fax phone number for the organization where this application or proceeding is assigned is 703-872-9306.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see <http://pair-direct.uspto.gov>. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).



Erik Kielin
Primary Examiner
June 26, 2005